

# Quality Control of Fortified Flour

## **Chemical Test: Understand your measurement method and your result**



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**Regional Workshop on QA/QC of Flour Fortification**

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# Outline

1. Measurement methods
2. Validation of methods
3. Measurement uncertainty
4. Case study A: vitamin A fortified flour
5. Case study B: iron fortified flour

# Measurement methods for micronutrients in food: *CODEX*

*Standard 234 – 1999 contain method amendments adopted 2014*

1. Fluorometry
2. Colorimetry
3. Spectrophotometry
4. Microbioassay
5. Rat bioassay
6. HPLC (added 2011)

*Accessed 2015*

# Recommended methods for specific applications

## IRON

- AAS (Atomic absorption spectroscopy)
- ICP-OES (Inductively coupled plasma atomic emission spectroscopy)
- Colorimetric assay

## VITAMIN A

- Colorimetric assay
- Fluorometry
- HPLC (high performance liquid chromatography)

## FOLIC ACID

- Microbiological assay
- Immunoassay
- Optical Biosensors

## IODINE

- ICP-OES
- Colorimetric assay
- Titration

# Each measurement method must be validated for a specific type of food and the type of micronutrient

## IRON

- NaFeEDTA
- Ferrous fumarate
- Ferrous sulfate
- Electrolytic iron
- Intrinsic iron

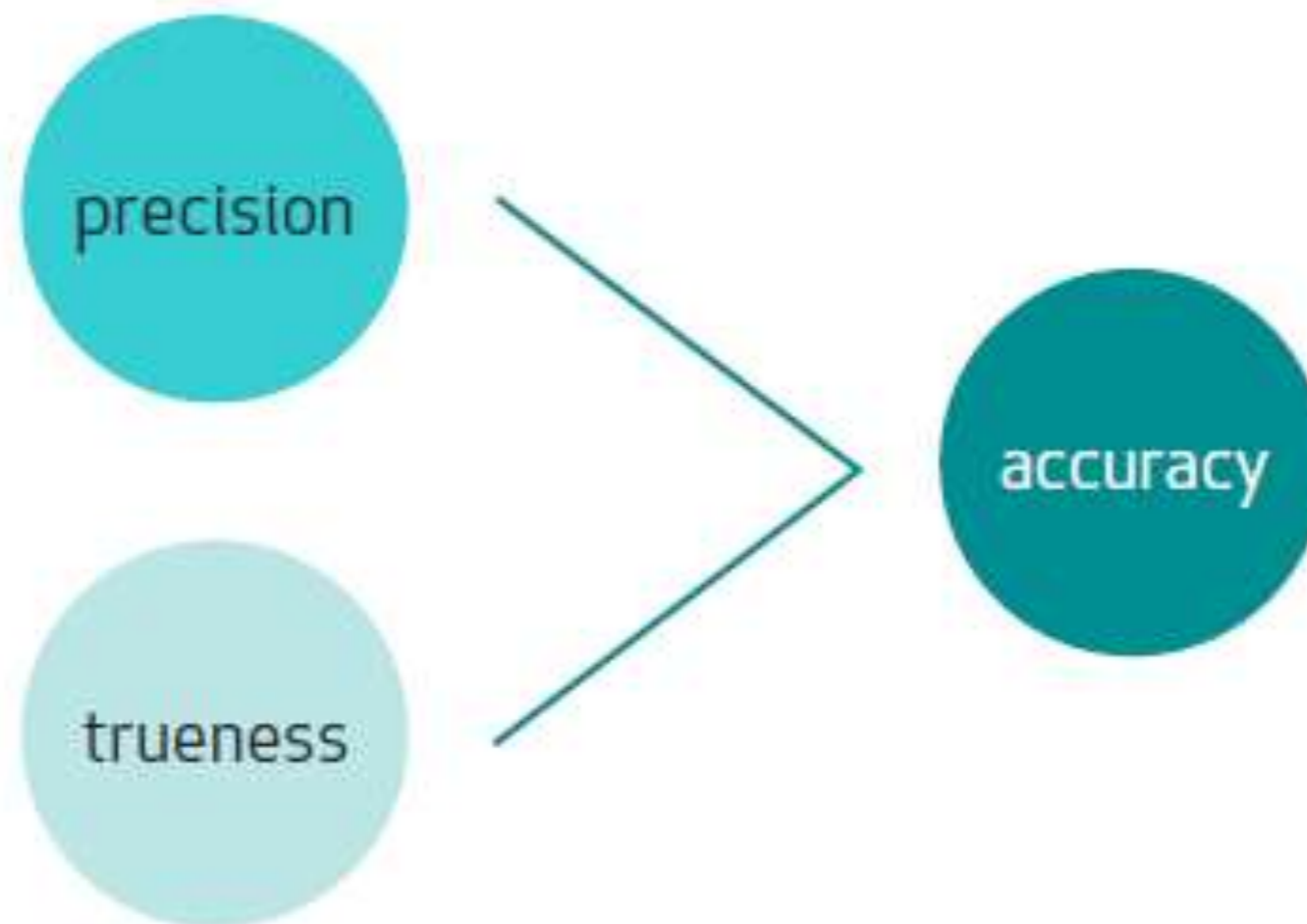


## VITAMIN A

- Vitamin A palmitate
- Vitamin A acetate
- Vitamin A propionate
- Encapsulated vitamin A



# Validation assesses the accuracy of the measurement



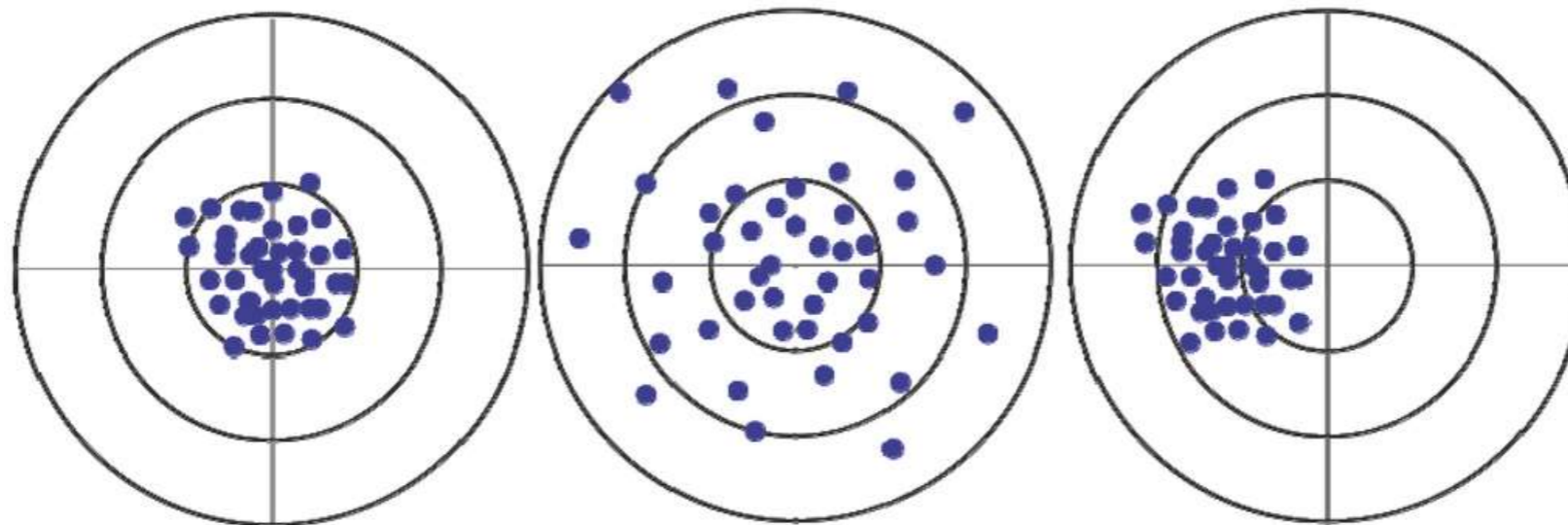


# International Standards Organization definition of accuracy

According to **ISO 5725-1**, **Accuracy** consists of Trueness and Precision

**Trueness** - proximity of measurement results to the true value

**Precision** - repeatability or reproducibility of the measurement



High trueness  
High precision

High trueness  
Low precision

Low trueness  
High precision

# PRECISION



# Multiple factors influence the precision of measurement



- the sample itself



- the analyst

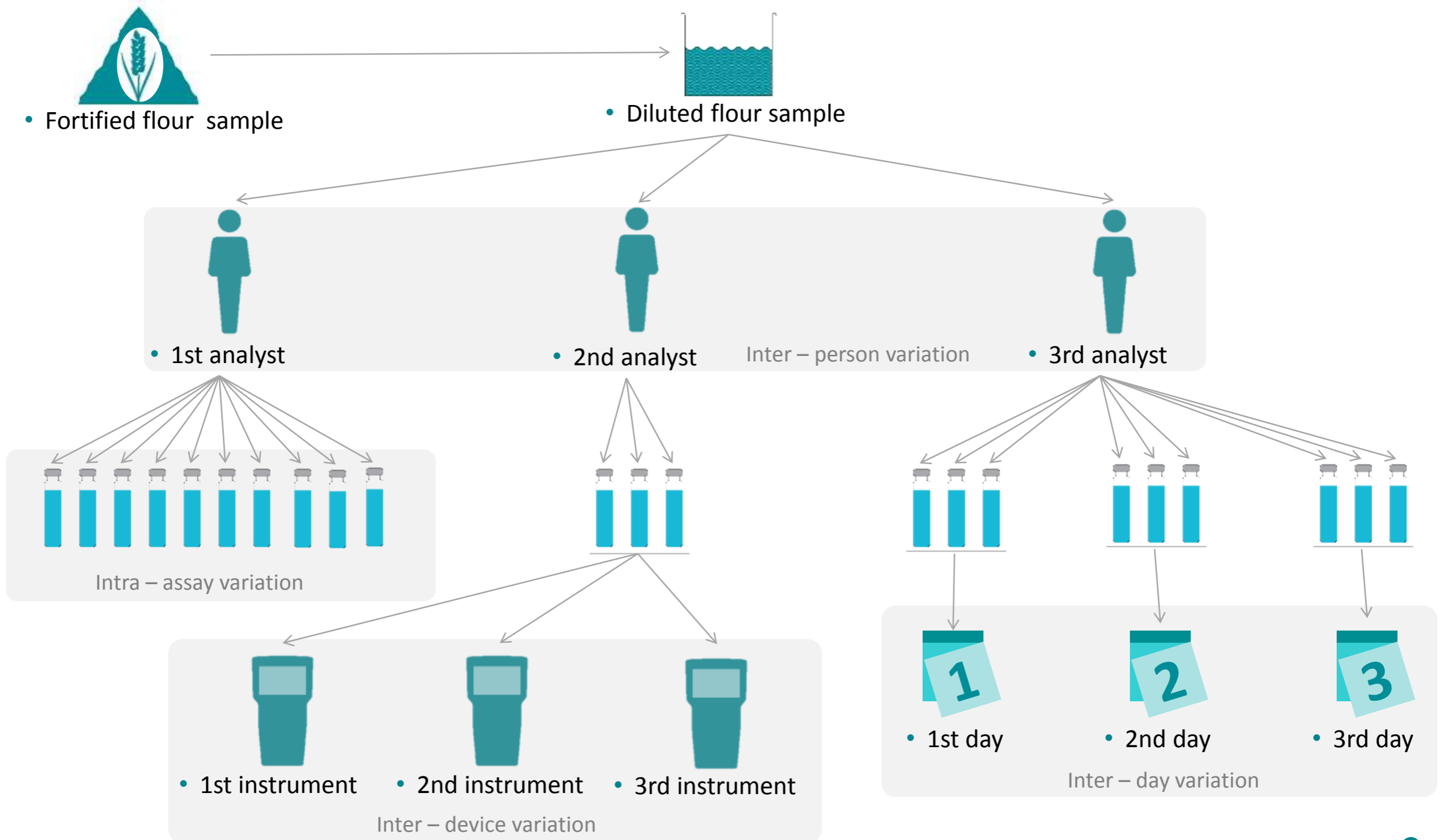


- the instrument



- the environment

# Validation protocol example for precision assessment



# Calculation of precision based on results of repeated measurements

$$\frac{\text{STANDARD DEVIATION}}{\text{MEAN}} \times 100\% =$$

= Coefficient of variation (CV) in %

Coefficient of Variation with 95% confidence level = 1.96 x CV

Example: (4 ppm/40 ppm) x 100% = 10%  
95% confidence = 1.96 x 10% = 19.6%

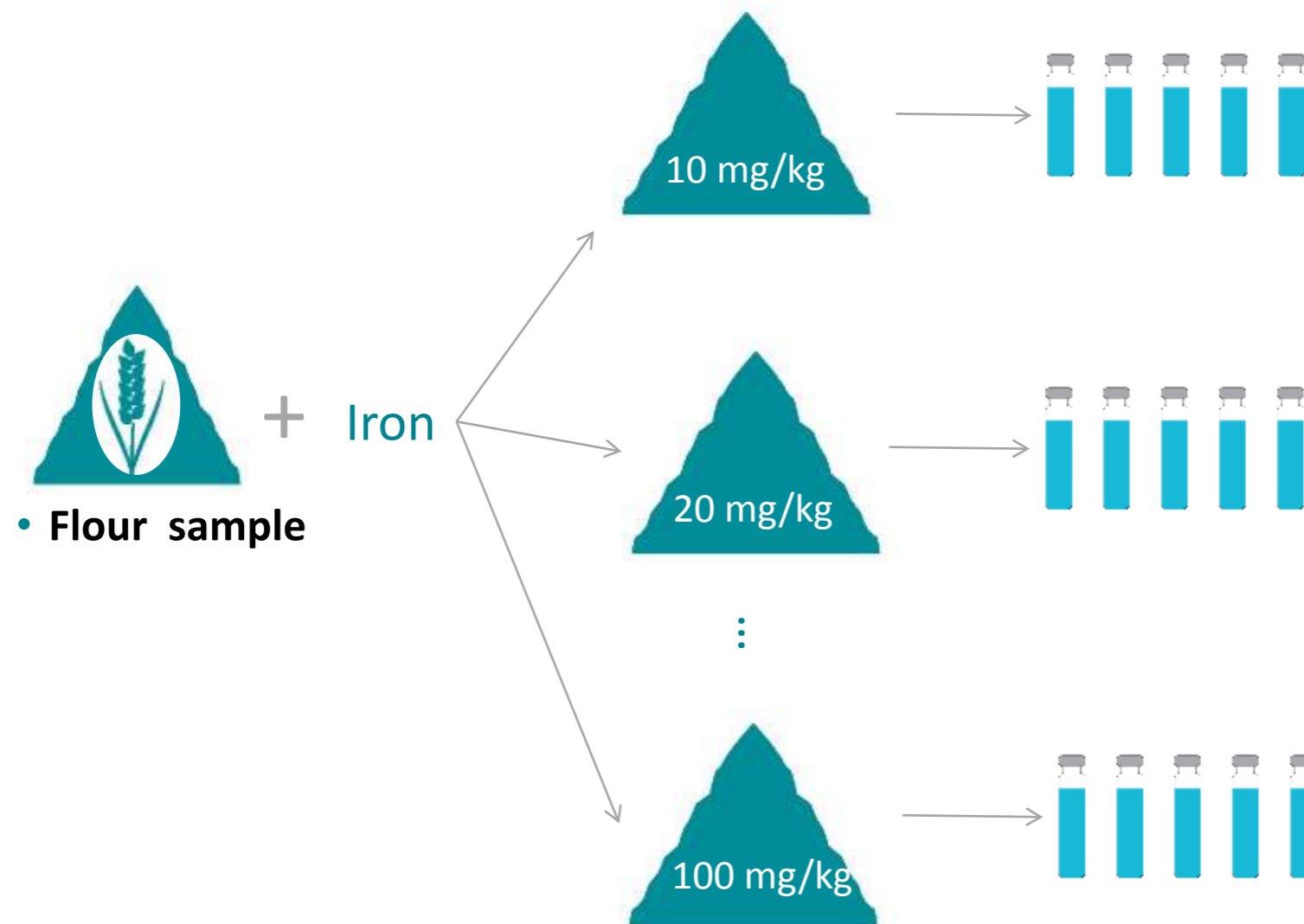
# AACC ring trial of 14 USA laboratories with 7 flour samples shows significant variation

Analyte	Minimum CV x 2	Maximum CV x 2	Mean CV x 2
Iron	12%	54%	<b>22%</b>
Zinc	9%	16%	<b>13%</b>
Vitamin A	63%	370%	<b>141%</b>
Folic Acid	30%	82%	<b>45%</b>

<http://www.aaccnet.org/checksample/>

# TRUENESS

# Validation protocol example for trueness assessment



# Calculation of trueness is based on the recovery of added analyte

$$\frac{\text{MEASURED AMOUNT}}{\text{ADDED AMOUNT}} \times 100\% = \text{Trueness in \%}$$

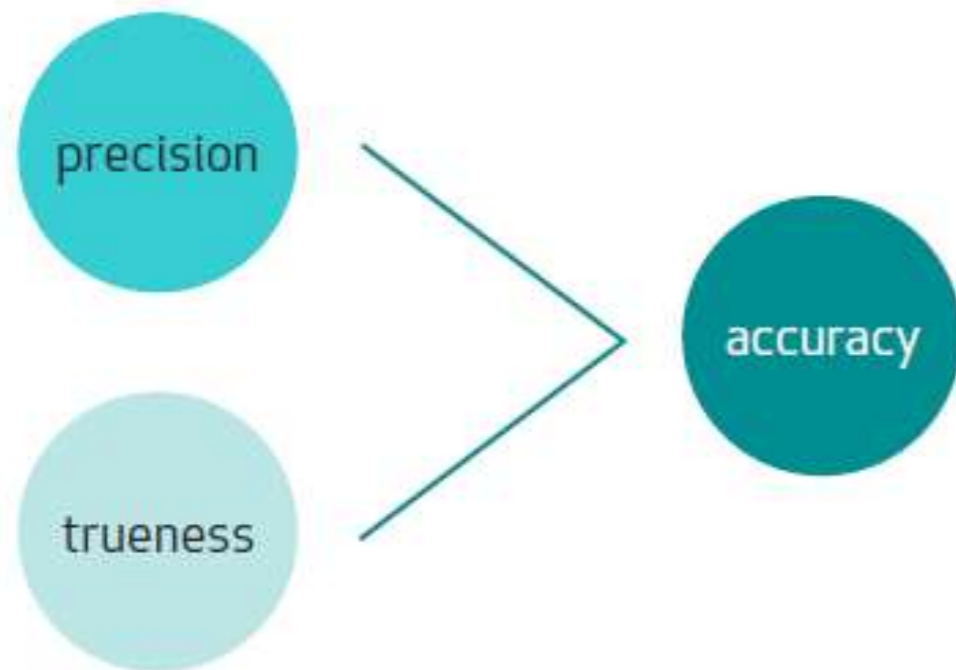
Trueness is expressed in terms of bias:  
how far is the result from the true value in %?

Example:  $(38 \text{ ppm}/40 \text{ ppm}) \times 100\% = 95\%$   
Bias:  $100\% - 95\% = 5\%$



# ACCURACY

# Accuracy = precision + trueness



- **Precision** is expressed in terms of coefficient of variation (**CV**) in % at 95% confidence level
- **Trueness** is expressed in terms of **bias**: how far is the result from the true value in %
- **Accuracy** is expressed in terms **Measurement uncertainty**, that combines CV and Bias.

# Calculation of measurement uncertainty to express the accuracy

$$\text{BIAS} + (1.96 \times \text{CV}) = \text{Measurement Uncertainty, \%}$$

$$\text{Example: } 5\% + (1.96 \times 10\%) = 24.6\%$$

# VERY IMPORTANT

## Sampling: Representative sample



→ Variation for single small samples can be as high as  $\pm 100\%$ .



**Pool/Combine multiple single samples in one composite sample**



→ Control Sample

# Conditions to be met for proper quality control

1. Measurement Uncertainty is Known
2. Sample is Representative
3. Standard is Set
4. Permitted Tolerance is Set

# Case Study A: Vitamin A in flour

**1. Measurement Uncertainty**

±20%

**2. Representative Sample**



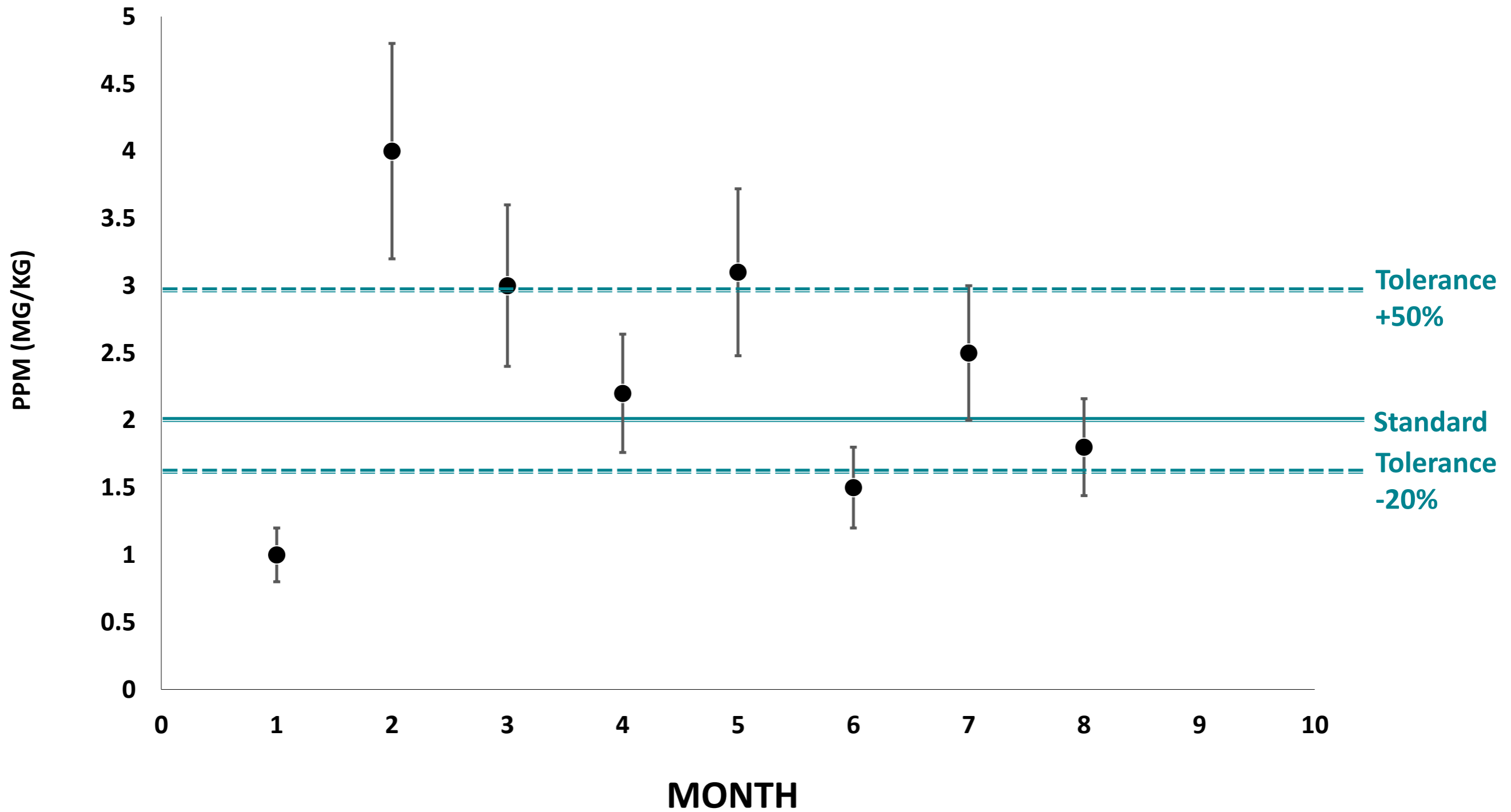
**3. Standard**

2 ppm

**4. Permitted Tolerance**

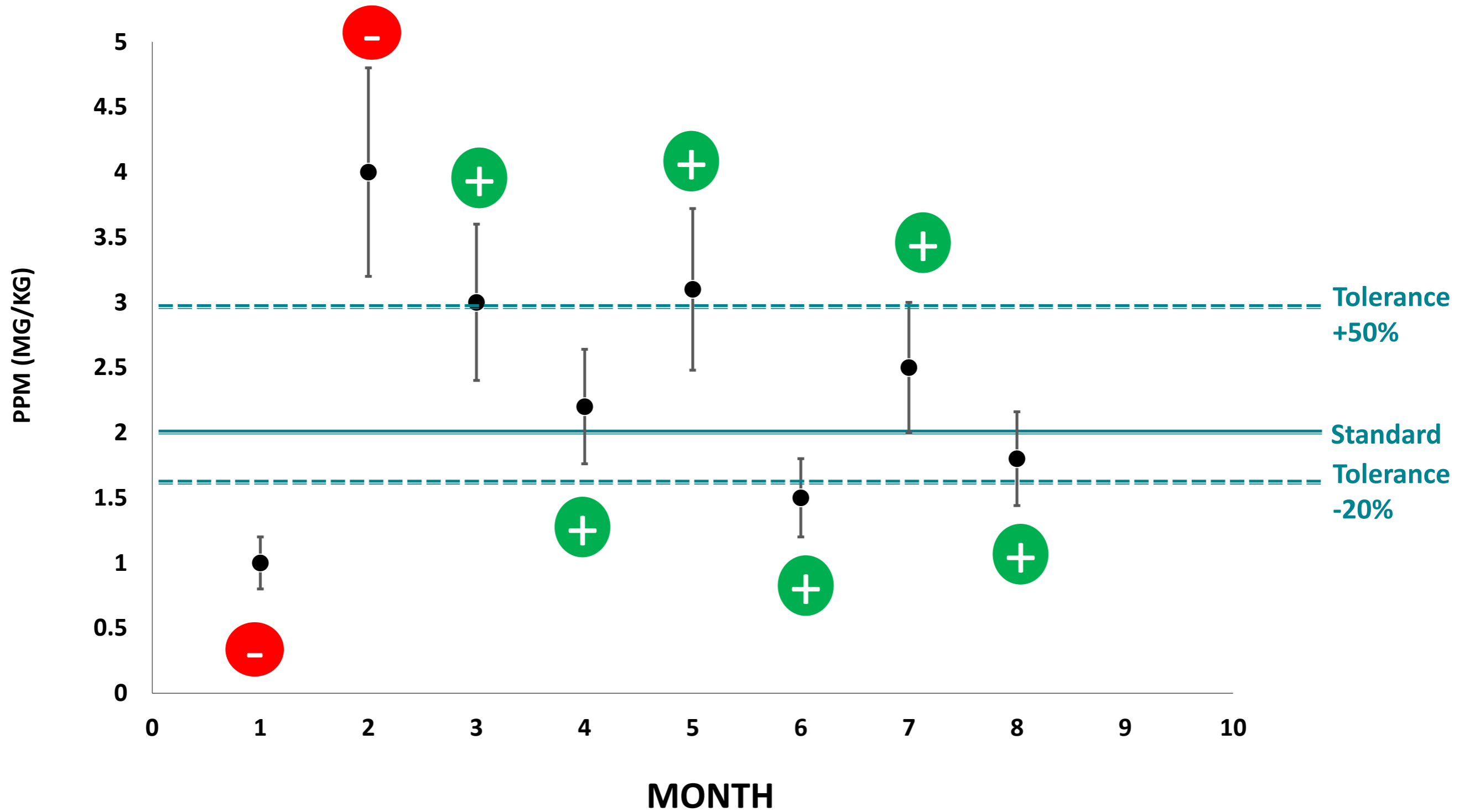
+50%  
-20%

# Case Study A: Vitamin A in flour





# Case Study A: Vitamin A in flour



# Case Study B: Iron in flour

**1. Measurement Uncertainty**

±25%

**2. Representative Sample**



**3. Standard**

20 ppm

**4. Permitted Tolerance**

±20%

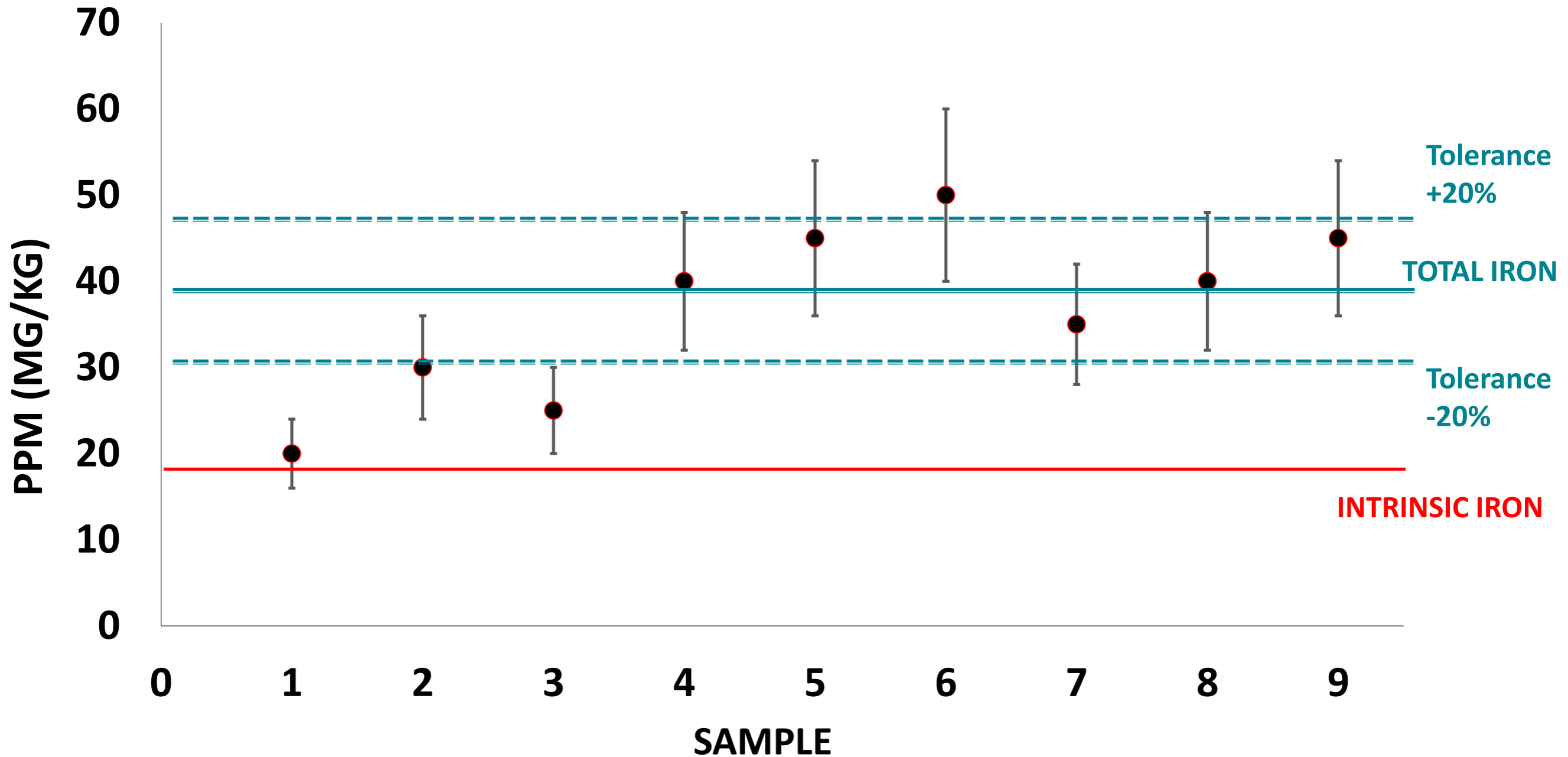
# Case Study B: Iron in flour

## Additional factor is Intrinsic iron

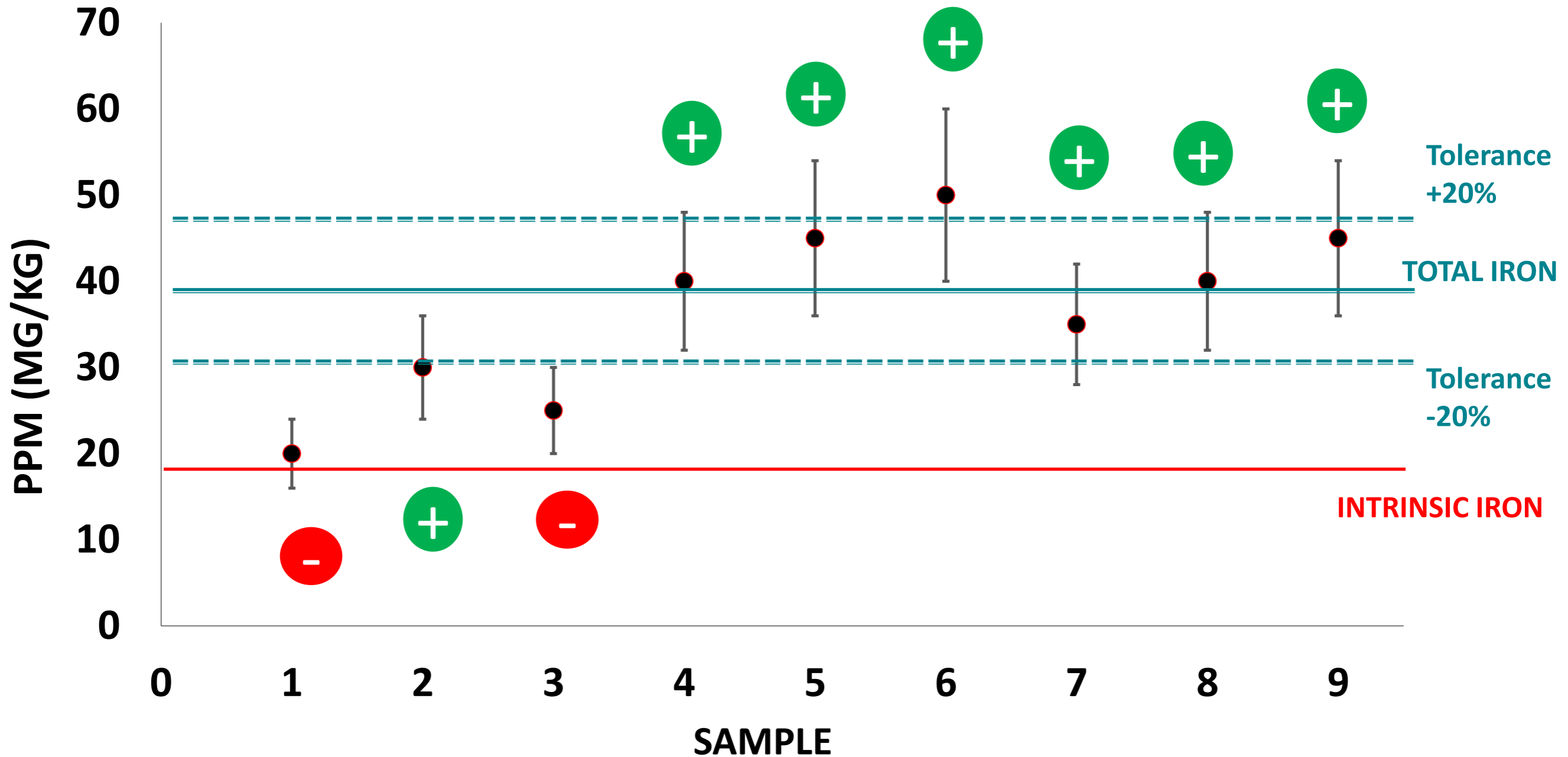
1. Self-rising white flour: 10-20 ppm
2. Bread flour: 20-40 ppm
3. Brown flour: 40-60 ppm



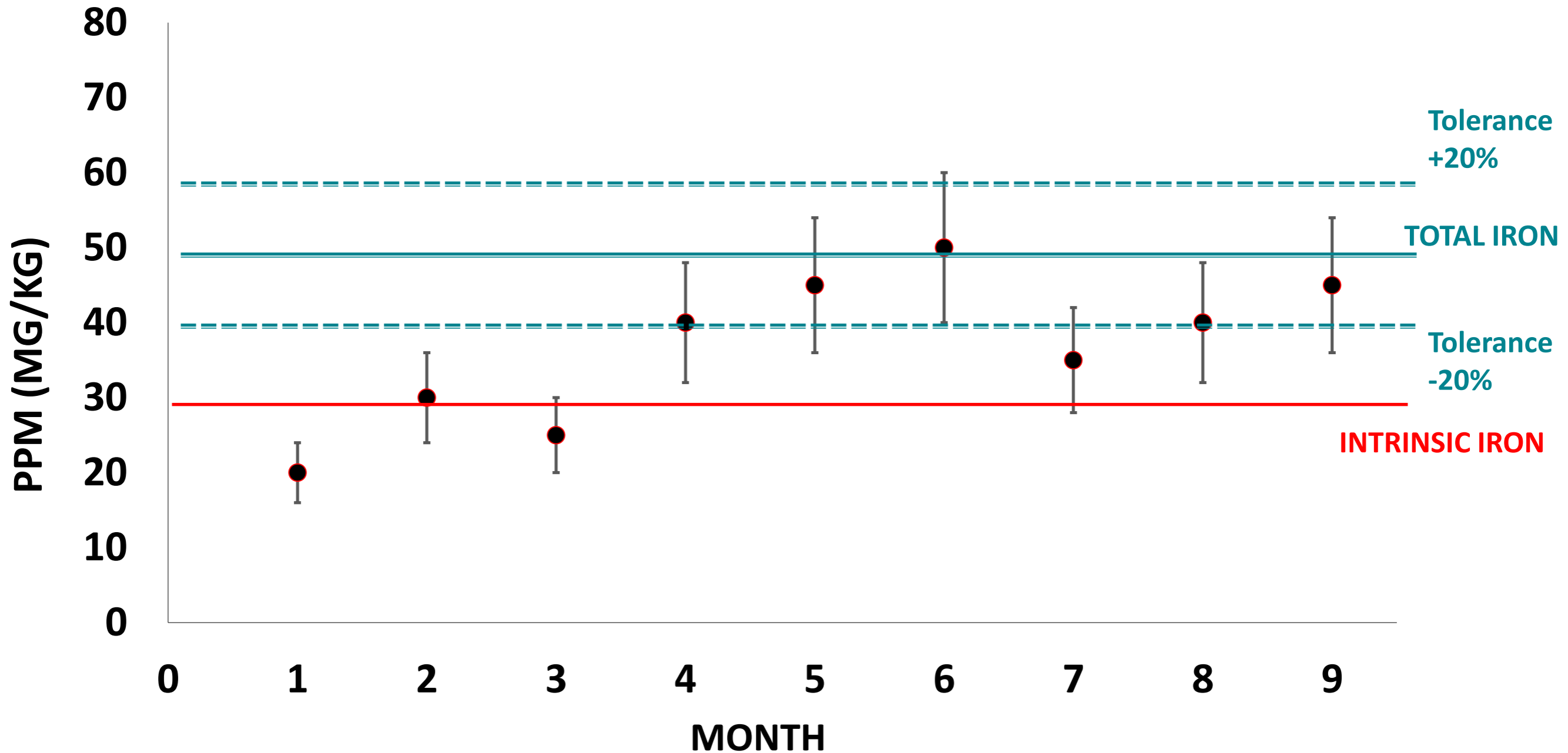
# Scenario 1: Self-rising flour with 20 ppm natural iron; 20 ppm added iron



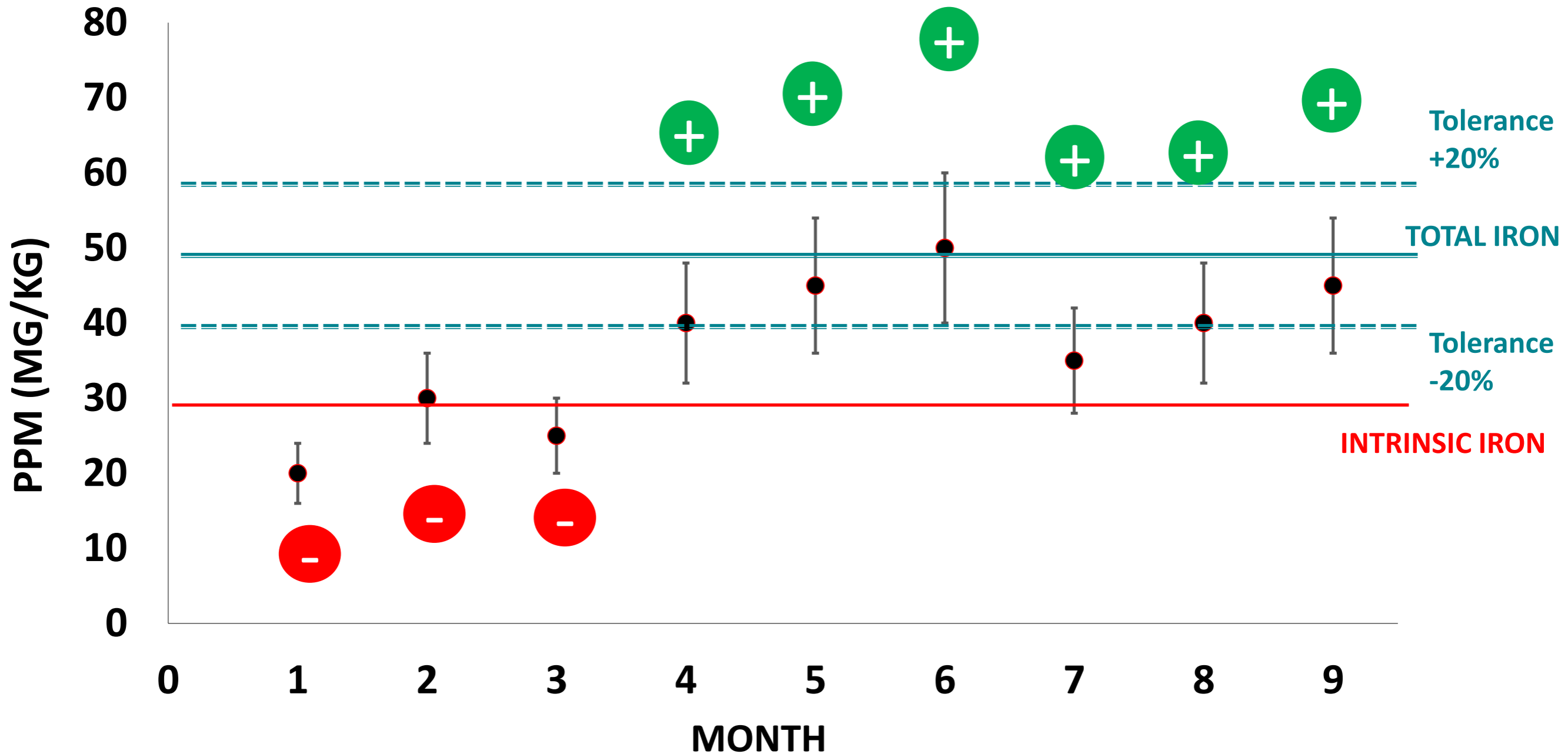
# Scenario 1: Self-rising flour with 20 ppm natural iron; 20 ppm added iron



# Scenario 2: Bread flour with 30 ppm natural iron; 20 ppm added iron



# Scenario 2: Bread flour with 30 ppm natural iron; 20 ppm added iron





# Case Study B: Iron in flour

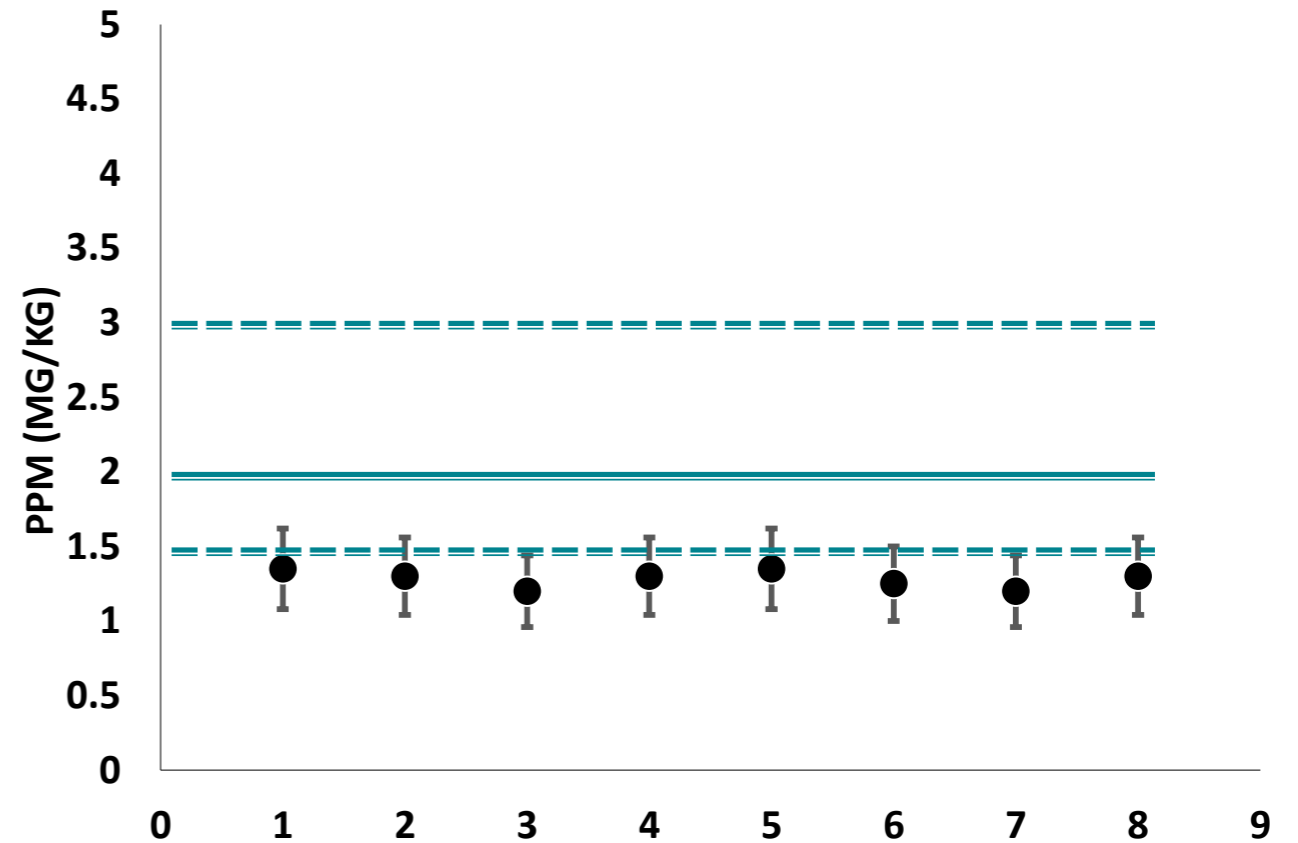
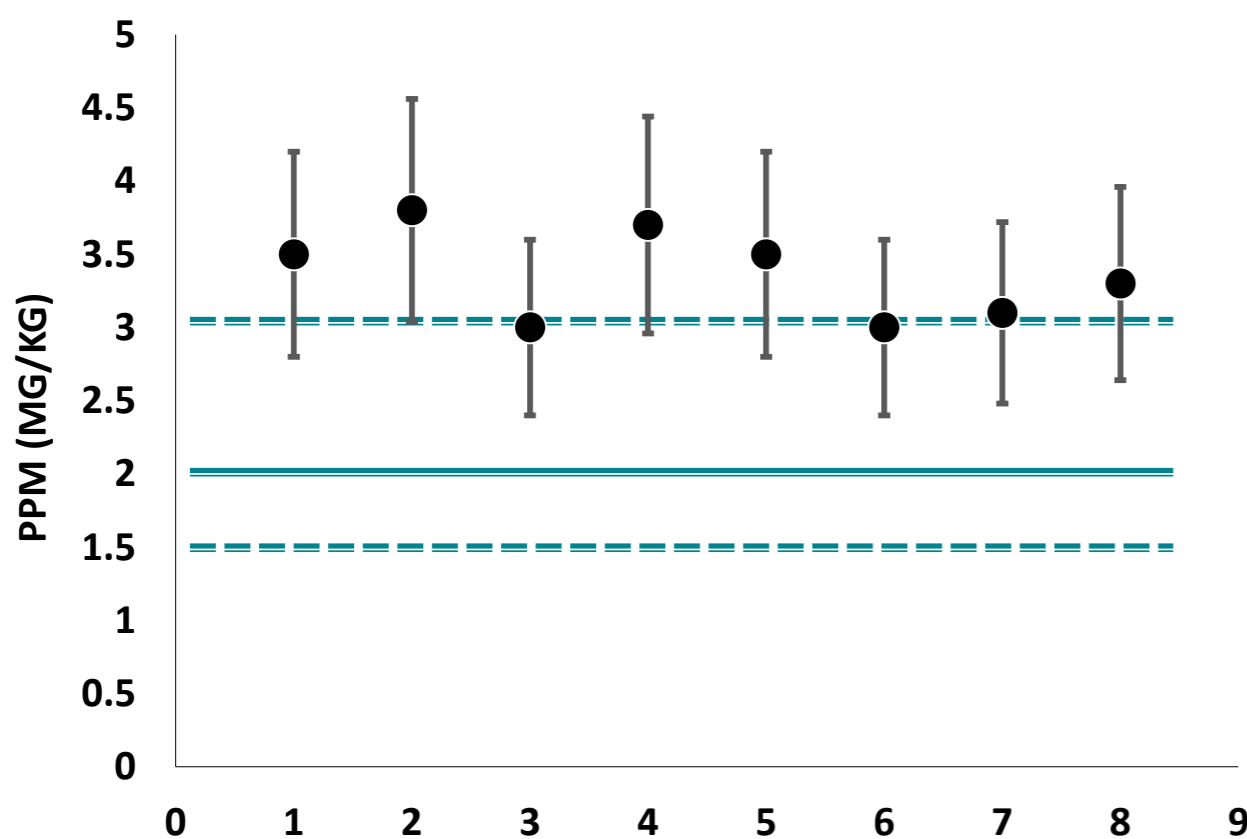
## Additional factor is Intrinsic iron

1. Self-rising white flour: 10-20 ppm
2. White bread flour: 20-40 ppm
3. Brown bread flour: 40-60 ppm



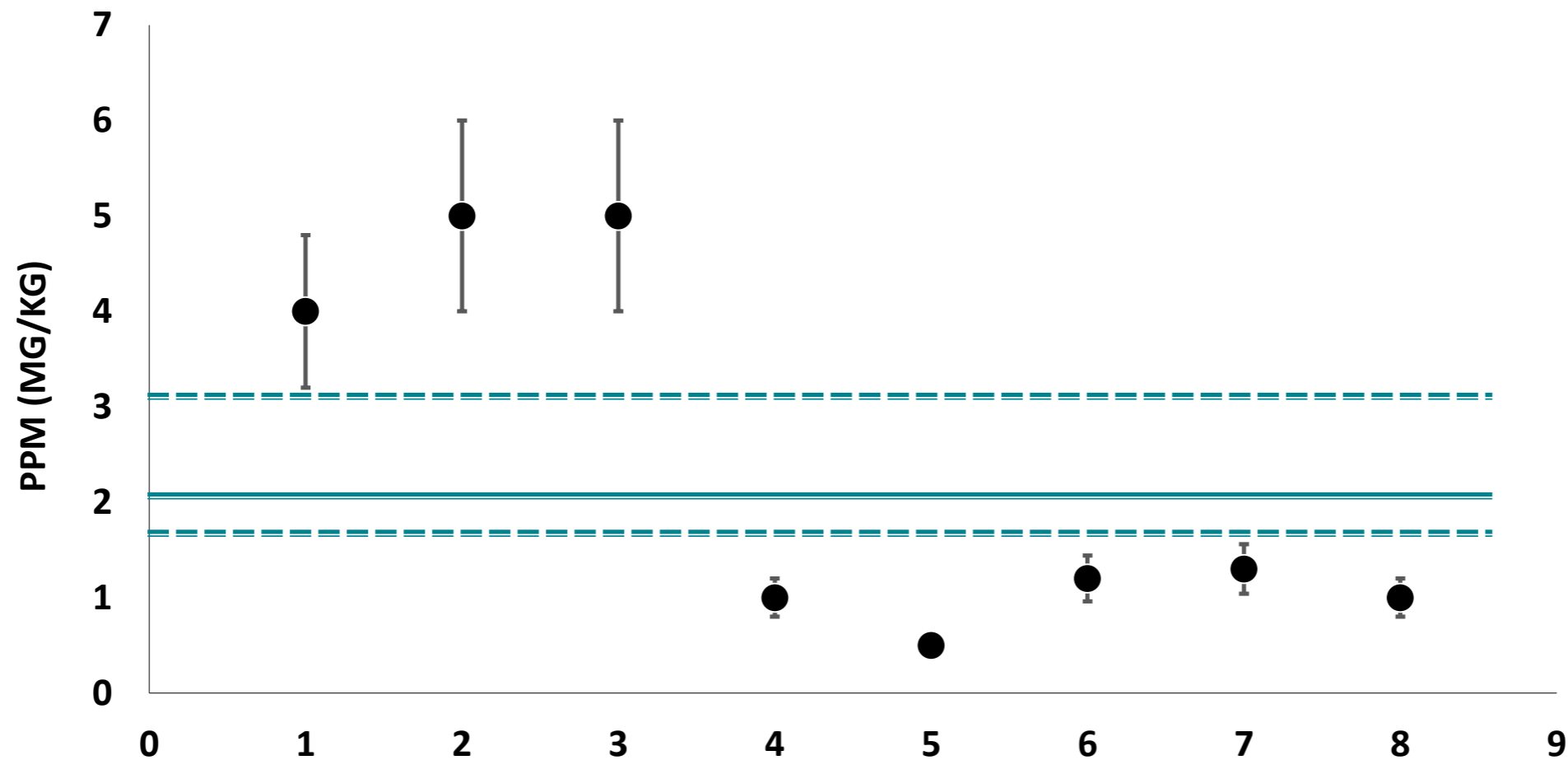
Legal definition of different flour types and implications for total iron standard and tolerance limits!!!

# Recommendation 1



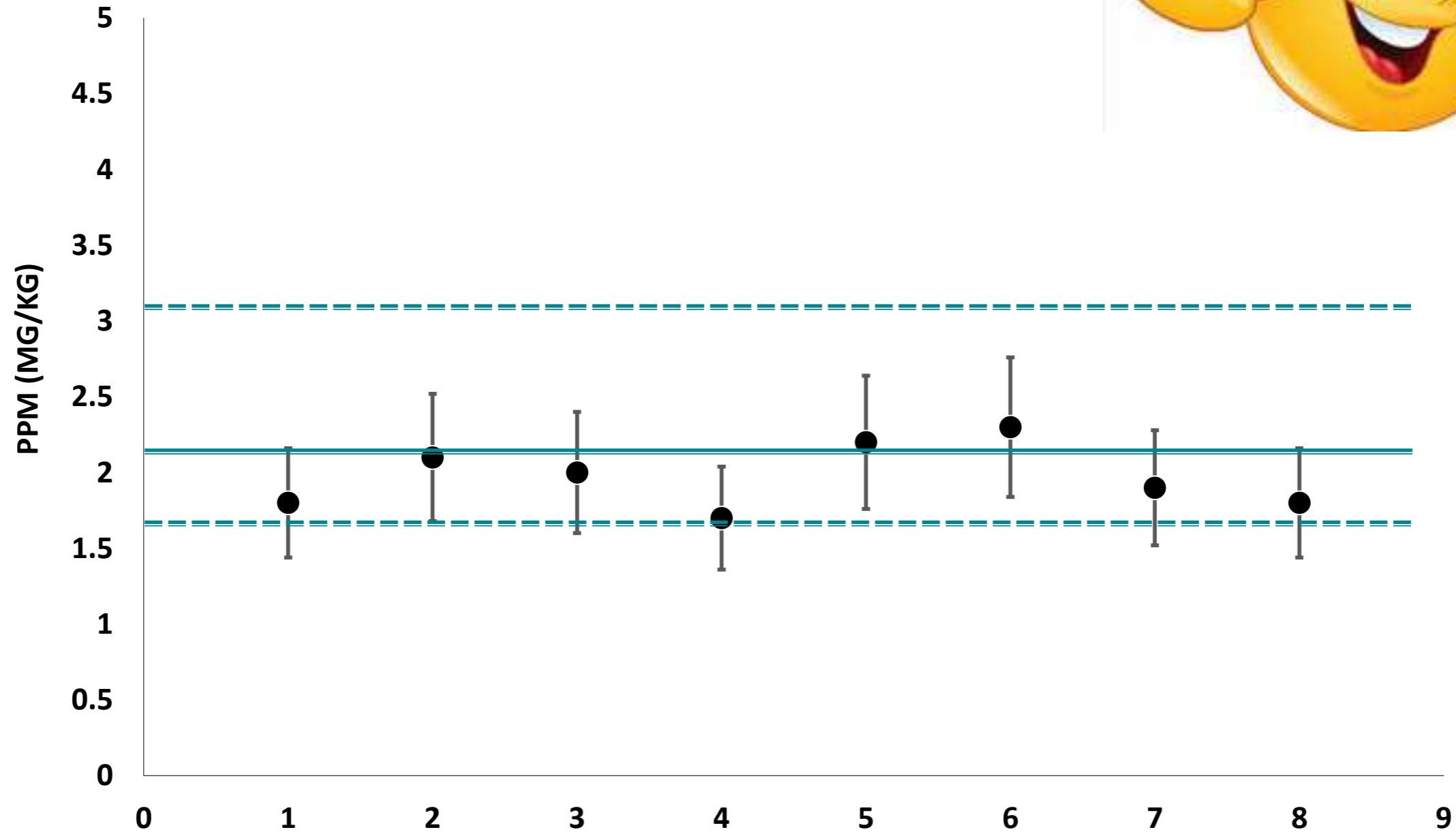
If the analysis results are repeatedly at the upper/lower limits of the permitted tolerance, the food business operator must adopt more effective in-house control and make the necessary changes in the production process or the labelling.

# Recommendation 2



If the indicated nutrient contents of the foodstuff deviate repeatedly from the set tolerance limits, taking the measurement uncertainty of the analysis method into account, the foodstuff is not acceptable and may not be kept for sale.

# Optimal Situation



# Conditions to be met for proper quality control

1. Measurement Uncertainty is Known
2. Sample is Representative
3. Standard is Set
4. Permitted Tolerance is Set

# THANK YOU!

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